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Draa Sfar, Morocco: A Visean (331 Ma) pyrrhotite-rich, polymetallic volcanogenic massive sulphide deposit in a Hercynian sediment-dominant terrane

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Abstract

Draa Sfar is a Visean, stratabound, volcanogenic massive sulphide ore deposit hosted by a Hercynian carbonaceous, black shale-rich succession of the Jebilet terrane, Morocco. The ore deposit contains 10 Mt grading 5.3 wt.% Zn, 2 wt.% Pb, and 0.3 wt.% Cu within two main massive sulphides orebodies, Tazakourt (Zn-rich) and Sidi M'Barek (Zn–Cu rich). Pyrrhotite is by far the dominant sulphide (70 to 95% of total sulphides), sphalerite is fairly abundant, chalcopyrite and galena are accessory, pyrite, arsenopyrite and bismuth minerals are rare. Pyrrhotite is monoclinic and mineralogical criteria indicate that it is of primary origin and not formed during metamorphism. Its composition is very homogeneous, close to Fe₇S₈, and its absolute magnetic susceptibility is 2.10^{−3} SI/g. Ar–Ar dating of hydrothermal sericites from a coherent rhyolite flow or dome within the immediate deposit footwall indicates an age of 331.7 ± 7.9 Ma for the Draa Sfar deposit and rhyolite volcanism.

The Draa Sfar deposit has undergone a low-grade regional metamorphic event that caused pervasive recrystallization, followed by a ductile–brittle deformation event that has locally imparted a mylonitic texture to the sulphides and, in part, is responsible for the elongated and sheet-like morphology of the sulphide orebodies. Lead isotope data fall into two compositional end-members. The least radiogenic end-member, (²⁰⁶Pb/²⁰⁴Pb = 18.28), is characteristic of the Tazakourt orebody, whereas the more radiogenic end-member (²⁰⁶Pb/²⁰⁴Pb ~ 18.80) is associated with the Sidi M'Barek orebody, giving a mixing trend between the two end-members. Lead isotope compositions at Draa Sfar testify to a significant continental crust source for the base metals, but are different than those of the Hajar and South Iberian Pyrite Belt VMS deposits.

The abundance of pyrrhotite *versus* pyrite in the orebodies is attributed to low *f*O₂ conditions and neither a high temperature nor a low aH₂S (below 10^{−3}) is required. The highly anoxic conditions required to stabilize pyrrhotite over pyrite are consistent with formation of the deposit within a restricted, sediment-starved, anoxic basin characterized by the deposition of carbonaceous, pelagic sediments along the flank of a rhyolitic flow-dome complex that was buried by pelitic sediments. Deposition of sulphides likely occurred at and below the seafloor within anoxic and carbonaceous muds.

Draa Sfar and other Moroccan volcanogenic massive sulphide deposits occur in an epicontinental volcanic domain within the outer zone of the Hercynian belt and formed within a sedimentary environment that has a high pelagic component. In spite of the diachronous emplacement between the IPB deposits (late Devonian to Viséan) and Moroccan deposits (Dinantian), all were formed around 340 ± 10 Ma following a major phase of the Devonian compression.

Keywords: Massive sulphide deposit; Pyrrhotite; Draa Sfar; Morocco; Hercynian

1. Introduction

Recent exploration conducted by REMINEX in the Jebilet and Guemassa terranes of Morocco, as well as the commencement of mining at the Draa Sfar deposit, has resulted in increased accessibility to the deposit and its host rocks allowing new data to be obtained from underground and surface mapping and from drill core. This new data for the Draa Sfar deposit has significant implications for all of the sparsely documented volcanogenic massive sulphide (VMS) deposits in the Jebilet and Guemassa terranes of Morocco.

A companion paper by Belkabir et al. (2008-this issue) focuses on the geology, structure, volcanic environment and hydrothermal alteration at Draa Sfar. In this paper we focus on the Draa Sfar sulphide deposit that ranks second for Zn production in Morocco. We also describe the geological, mineralogical and lead isotopic characteristics of the sulphide deposit and propose a new model that may be applicable to other VMS deposits in the Jebilet and Guemassa terranes.

With production at Draa Sfar just starting in September 2005, underground workings, drill core and surface outcrops provided excellent exposures of the deposit and of the volcano-sedimentary lithofacies that underlie, host and overlie it. This access has allowed us to collect representative samples of the Draa Sfar orebodies as well as the host volcanic-sedimentary succession.

Drill cores through the orebody have been analyzed for major and trace elements in order to characterize the ores and to determine the distribution of base and precious metals. The ore mineralogy has been determined through detailed petrographic analysis of 50 samples, and X-ray diffractometry (XRD), scanning electron microscopy (SEM), and electron probe microanalysis (EPMA) of these samples. Fifteen lead isotope analyses were obtained on representative samples of the various ore facies in order to better constrain the source of metals and genesis of the massive sulphide deposit. Preliminary magnetic susceptibility measurements on the massive sulphide orebodies were also made.

2. Geological framework

The Draa Sfar mine is located at the southern limit of the Jebilet massif, in the southern part of the West Moroccan Meseta (Fig. 1). The Jebilet massif, extending 170 km EW and 40 NS, is interpreted to be an intracontinental, Devonian–Carboniferous rift-basin that formed during a period of extension during rifting of continental crust in a back-arc geodynamic environment ([Huvelin, 1977] and [Aarab and Beauchamp, 1987]). The Jebilet massif consists of three lithotectonic domains, the Western, Central, and Eastern Domains (Fig. 1b), which are separated by major shear zones ([Huvelin, 1977] and [Lagarde and Choukroune, 1982]). The Western Domain consists of Cambrian and Ordovician rocks, whereas the Central

domain consists of metamorphosed and deformed upper Viséan (350 to 340 Ma) rocks. The Eastern Domain is also comprised of Viséan rocks, but is covered by nappes that contain strata ranging from Ordovician to early Carboniferous in age.

The Central Jebilet Domain (CJD) consists of a thick succession of argillites and carbonaceous argillites (up to 1.5 km thick), representing the lower part of the succession. These are overlain by Carboniferous, carbonaceous argillites and limestones (< 100 m) that define the top of an interpreted basin-fill succession (Bordonaro, 1983). The argillites of the CJD are interpreted by Beauchamp et al. (1991) to have been deposited in an anoxic, shallower water environment. The end of the Carboniferous is characterized by the development of the Mesetian basin where intense volcanic and volcano-sedimentary activity ([Bordonaro, 1983], [Aarab and Beauchamp, 1987] and [Bernard et al., 1988]) is characterized by three types of magmatism: (1) bimodal, subvolcanic rift-related tholeiitic gabbro and minor rhyolite and rhyodacite flows; (2) syntectonic calc-alkaline granite intrusions and (3) post-tectonic, microdiorite dykes and sills ([Huvelin, 1977], [Bernard et al., 1988] and [Essaifi et al., 2004]).

Deformation in the CJD is marked by the Hercynian east–west D_1 shortening that is responsible for the orientation of the north-striking, transposed strata, development of the main regional S_1 schistosity (striking NS and subvertical), regional folds that trend 030° , and the emplacement of syntectonic granites. The D_1 deformation is associated with low-grade regional metamorphism, typified by a quartz–muscovite–chlorite assemblage. D_1 shortening was followed by an N–S D_2 shortening (S_2 crenulation cleavage) and late brittle deformation ([Huvelin, 1977], [Bordonaro, 1983] and [Bernard et al., 1988]).

Several massive sulphide deposits, including Draa Sfar, are hosted by the volcano-sedimentary succession within the CJD (Fig. 1b). The deposits are stratiform, characterized by a sulphide mineralogy dominated by pyrrhotite (70 to 95% of sulphide minerals) and range from Zn-rich polymetallic mineralization (*e.g.*, Koudiat Aicha — 4.5 Mt grading 2.7 wt.% Zn, 1.2 wt.% Pb and 0.6 wt.% Cu, Draa Sfar — 10 Mt grading 5.3 wt.% Zn, 2 wt.% Pb and 0.3 wt.% Cu), to low-grade sulphide deposits (*e.g.*, Kettara; 30 Mt of pyrrhotite at 0.7 wt.% Cu) (Table 1). The sulphide deposits are pre-tectonic as they are affected by the S_1 schistosity and generally occur as folded and dislocated sheet-like orebodies ([Bordonaro, 1983] and [Bernard et al., 1988]).

3. Geology of the Draa Sfar Deposit

A thorough description of the lithofacies, structure and stratigraphy of the Draa Sfar deposit is provided by Belkabir et al. (2008-this issue) and only a summary is provided below. The Draa Sfar deposit occurs within a dominantly sedimentary succession characteristic of the lower CJD strata. The orebodies occur in the upper part of the Sarhlef series and are underlain by more than 500 m of argillite and carbonaceous argillite with intercalated siltstone and a subordinate, but locally dominant coherent rhyolite and volcanoclastic lithofacies (Fig. 2 and Fig. 3). The Draa Sfar deposit is conformably overlain by a thinly bedded carbonaceous argillite and intercalated siltstone that is intruded by a thick gabbro sill. The host rocks strike NNE–SSW, and are flattened parallel to the S_1 foliation (Fig. 2). Within the least-deformed volcanic and sedimentary rocks, younging directions were determined from primary structures (crossbedding, grading, syndimentary loading structures), and these consistently indicate that the strata immediately above and below the deposit are overturned and young to the west (Fig. 2). However, on a slightly larger scale, facies variations (*e.g.*, coherent rhyolite to

volcaniclastic facies) and facing directions indicate that the rocks are folded about a north-trending F_1 anticline that places the Draa Sfar deposit on the west limb of this fold (Fig. 2 and Fig. 3). The rocks contain assemblages of chlorite, muscovite, quartz, albite, and rare oligoclase–orthoclase indicative of a greenschist metamorphic grade.

3.1. The Draa Sfar deposit: morphology and characteristics

The Draa Sfar deposit contains 10 Mt of ore grading 5.3 wt.% Zn, 2 wt.% Pb, 0.3 wt.% Cu, which places it in the Zn–Pb–Cu group of volcanogenic massive sulphide (VMS) deposits, close to the Zn–Pb tie-line of Franklin et al. (1981). However, the dominance of pelitic sedimentary rocks in both the stratigraphic hanging wall and footwall places the Draa Sfar deposit in the pelitic subgroup of the siliciclastic–felsic type of Franklin et al. (2005), which also includes VMS deposit of the Iberian Pyrite Belt, Spain and Portugal, and the Bathurst mining Camp, Canada.

The Draa Sfar deposit strikes N to NNE and extends for more than 1.7 km. The deposit is overturned and dips steeply to the east and faces west. It contains two orebodies: a southern, zinc-rich orebody called Tazakourt, and a northern more copper-rich orebody called Sidi M'Barek that is dismembered by cross-faults (Fig. 2). At the surface the two orebodies are separated by the Tensift River, but there is no structural discontinuity between the orebodies. The two orebodies consist of Fe–Zn–Cu–Pb lenses of massive sulphides that are predominantly composed of pyrrhotite.

Tazakourt consists of one lens of massive sulphide that extends N–S for more than 1 km. The footwall to Tazakourt consists of felsic volcanic rocks, argillite and carbonaceous argillite, interbedded with siltstone, whereas the hanging wall consists of carbonaceous argillite and interbedded siltstone (Fig. 3 and Fig. 4). The massive sulphides are hosted by argillite and interbedded siltstone (Fig. 4 and Fig. 5). The orebody has been delineated between the – 50 m and – 900 m levels, and remains open at depth. The Tazakourt orebody, which lies stratigraphically above a coherent rhyolite, is the main Zn-rich massive sulphide lens with an average grade of 8 to 9 wt.% Zn, 0.3 wt.% Cu and 1 to 2 wt.% Pb (Fig. 2). Base metal zoning (Zn relative to Cu) in this body indicates a trend to Cu-enrichment in the south and Zn-enrichment to the north. Tazakourt is well delimited by drill-holes and represents the main ore lens at Draa Sfar.

The Sidi M'Barek orebody consists of two massive sulphides lenses that are also hosted by the argillite–siltstone lithofacies (Fig. 3 and Fig. 5). The western or upper Zn-rich lens (6 to 7 wt.% Zn and ~ 0.3 wt.% Cu, < 1 wt.% Pb) is interpreted to be a continuation of the Tazakourt ore lens. The eastern or lower lens is more Cu-rich (2 to 3 wt.% Cu, < 1 wt.% Zn, < 1 wt.% Pb and ~ 600 to 700 ppm Co). Sidi M'Barek is only exposed in drill cores and has not yet been delineated. Drilling indicates that both the upper and lower lenses are vertically continuous but are discontinuous along strike (less than 70 m in along-strike continuity).

The pyrrhotite-rich orebodies occur on the east limb of a large anticline that plunges towards north ([Bernard et al., 1988] and [Belkabir et al., 2008-this issue]) (Fig. 2). The Tazakourt orebody is commonly boudinaged with thicknesses that range from 0.2 to 20 m. Such variations in thickness are observed in plan and section from surface to the – 800m level and impart a “pinch and swell geometry” to the orebody (Fig. 3 and Fig. 4). Inclusions of chloritized argillite–siltstone occur within the massive sulphide lens and locally form piercement structures within massive pyrrhotite orebody (Belkabir et al., 2008-this issue). The

development of boudins and piercement structures reflects the difference in competency between the more competent pyrrhotite-rich ore lens and the less competent chloritized and schistose sedimentary rocks. The contact between the sulphide lens and the hanging wall carbonaceous argillites is an N10–70°E trending fault defined by tectonic breccias and fault gouge. The thickest sections of massive sulphide contain tectonic breccia and cm-scale (0.2 to 1 m long) lozenges of chloritized sedimentary rocks now altered to a talc–chlorite–carbonate assemblage or massive chloritized argillite that are cross-cut by post-S₁, barren pyrite veinlets (Fig. 6a,b).

Where the ore lens is thin (1 to 5 m) the sulphides are typically banded and may grade into zones of disseminated sulphides. Within the massive sulphide lens primary textures are typically not preserved. The footwall contact of the ore lens is often marked by pyrrhotite veinlets in chloritized argillite (Fig. 6h), by highly deformed massive bodies of black chlorite that is interpreted to replace argillite, or by pyrrhotite–chalcopyrite nodules disseminated within argillites (Fig. 6i; Belkadir et al., 2008-this issue).

A feeder or stockwork sulphide zone has not been identified at Draa Sfar, but the lower or eastern Cu-rich lens at Sidi M'Barek has a higher copper content and contains bismuth selenides that are characteristic of sulphide stockwork zones that underlie VMS deposits of the Iberian Pyrite Belt ([Marcoux et al., 1996] and [Leistel et al., 1998]). Bismuth selenides suggest that the lower, Cu-rich, ore lens at Sidi M'Barek may represent the upper part of a stockwork zone for the Draa Sfar ore deposit that has been strongly transposed by deformation. The more Cu-rich southern portion of the Tazakourt orebody may suggest proximity to another, yet undiscovered, stockwork zone?

Intense surface oxidation zones are known around Draa Sfar. At Tazakourt, there are surface oxidation zones related spatially to faults and fractures that intersect the massive sulphide lens but these are not well developed. However, at Sidi M'Barek a well-developed gossan extends to a depth of 20 m.

3.2. Mineralogy and textures of sulphides lenses

The Draa Sfar orebodies consist dominantly of pyrrhotite (70 to 95 vol.% of sulphides, but commonly up to 90 to 95 vol.% in Zn and Cu-depleted zones), with lesser sphalerite (1 to 10 vol.%), galena (0.5 to 5 vol.%) and chalcopyrite (1 to 5 vol.%, with the exception of the Cu-rich lower lens at Sidi M'Barek where chalcopyrite content reaches 15 vol.%), and with local concentrations of deformed pyrite (2 to 3 vol.% of total sulphides). Faint, but recognizable tectonic banding with massive sulphide reflects variations in the proportions of the different sulphide minerals. Accessory minerals include chlorite (clinochlore), siderite, talc, quartz, muscovite, and commonly ilmenite.

Pyrrhotite is always the dominant sulphide (Fig. 6 and Fig. 7). It is not nickel bearing (Ni contents below 150 ppm based on EPMA analyses) and pentlandite has only been observed rarely as minute flames (30 µm × 3–5 µm) in pyrrhotite from the Tazakourt orebody. EPMA indicate that pyrrhotite has a very homogeneous composition throughout the entire deposit, with a structural formula close to Fe₇S₈ and X_{FeS} close to 0.93 (Table 2). Three trace elements have been detected in the pyrrhotite: arsenic and bismuth, that occurs in pyrrhotite from all the sulphide lenses, but in various amounts (average grades of 600 and 500 ppm, respectively), and cobalt which is restricted to pyrrhotite in the northern, Cu-rich lower lens at Sidi M'Barek (900 ppm).

Sphalerite is the next most common sulphide and it can be the dominant sulphide within millimetre-wide bands (Fig. 7a–c). It is always associated with pyrrhotite. EPMA analyses indicate a uniform iron content for sphalerite that ranges from 6.0 to 8.2 wt.% Fe (9.4 to 12.8, mean 11.4 to mol% FeS) throughout the deposit (Table 2, Fig. 8). Cadmium is present, but in highly variable amounts (mean 500 ppm). Copper is also highly variable (0 to 2500 ppm) and higher values may be due to minute inclusions of chalcopyrite. Silver has not been detected and chalcopyrite dissolution within sphalerite has not been observed.

Galena occurs preferentially with sphalerite but is less abundant and Ag-free (Fig. 7b). Inclusions of silver minerals have not been observed. Chalcopyrite is of low abundance except within the lower or eastern lens at Sidi M'Barek, where it locally represents 10 to 15 vol.% of the ore and is the most abundant sulphide after pyrrhotite (Fig. 6 and Fig. 9).

Pyrite and marcasite are rare (< 1 vol.% of sulphides) at Tazakourt but are more common at Sidi M'Barek (2 to 3 vol.%). Pyrite shows two associations; 1) the first association is least common (Fig. 6 and Fig. 7), synchronous with pyrrhotite, and displays framboidal to ovoid textures that are usually considered as indicators of a primary, sedimentary origin; 2) the second association is much more common and is interpreted to be secondary, a product of the oxidative alteration of pyrrhotite (Fig. 6 and Fig. 7). It occurs as inclusions, sometimes a cm in size, within pyrrhotite, where it is associated with secondary magnetite and marcasite. Typically, the abundance of secondary pyrite is associated with carbonate veins that have a marcasite rim (Fig. 7f). The transformation of pyrrhotite to pyrite results in a volume loss and zones with secondary pyrite often appear vuggy and porous. Secondary pyrite also occurs as thin bands parallel to pyrrhotite-rich bands.

Arsenopyrite is the most common of the minor minerals. It is interpreted to have formed early as it always appears as fragmented or corroded crystals within other sulphides (Fig. 7a, c). Arsenopyrite is absent in the lower Cu-rich lens at Sidi M'Barek, although it is fairly common in zinc-rich lens at Tazakourt. However, Co-bearing arsenopyrite crystals of less than 0.1 mm in size have been reported from Sidi M'Barek (Barrakad et al., 1977).

EPMA analyses indicate that, regardless of grain size, arsenopyrite is As-rich (31.2 to 34.3 atom.% As and an average 33.26 atom.% As) (Table 2). Cobalt and selenium are present but in variable amounts: 0 to 0.47% (average 500 ppm) and 0 to 0.34% (average 700 ppm) respectively, but antimony and nickel were not detected.

A Bi–Se association was recognized in samples from the lower Cu-rich lens at Sidi M'Barek. Here, small (15 to 30 μm) inclusions of laitarite, $\text{Bi}_4(\text{Se},\text{S})_3$, and native bismuth were observed within chalcopyrite (Table 2). However, small (20 μm) paraganajuatite (Bi_2S_3), cobaltite (CoAsS) and mackinawite (FeS) inclusions reported from this orebody, and electrum, cassiterite and bismuthinite (Bi_2S_3) reported from Tazakourt, by Barrakad et al. (1977), were not observed in our samples.

Penetrative deformation and regional metamorphism have destroyed evidence of primary features within the sulphide lenses and obscured the recognition of multiple stages of mineralization with distinct assemblages. The exception to this is the obviously secondary association of pyrite–marcasite–magnetite, which is clearly late and related to a later fracturing and veining of the massive sulphide lenses.

The Draa Sfar ore mineralogy is similar to, but even more pyrrhotite-rich, than that of the Hajar deposit ([Ed Debi et al., 1998] and [Hibti et al., 1999]). This dominance of pyrrhotite is typical of Moroccan VMS deposits and is discussed below.

3.3. Textures and deformation of the ore

The ore textures at Draa Sfar are a product of several deformation events that have strongly modified the original textures and mineral assemblages. Regional metamorphism has caused an intense recrystallization of the sulphides, particularly in the Cu-rich lens of Sidi M'Barek (Fig. 6a). Pyrrhotite grains are equigranular, sub-mm in size, contiguous, with a polygonal texture and 120 °C angles, features that are indicative of metamorphic recrystallization (Fig. 9a, b), but which can form under low-grade metamorphic conditions (Gilligan and Marshall, 1987). Coexisting chalcopyrite occurs interstitial to pyrrhotite where it forms large anhedral, undeformed grains that display linear twinning. Where aligned, the chalcopyrite grains define a transposed S_1 foliation (Fig. 6a).

Following metamorphic recrystallization, the Draa Sfar deposit has undergone subsequent deformation events that have affected most of the ore lenses. There is abundant evidence for a later ductile deformation event responsible for the development of a mylonitic microfabric and tectonic banding within massive sulphide (Fig. 6 and Fig. 9). This mylonitic microfabric is characterized by the development of S_1 and S_2 foliations (C, C' and S planes), marked by stretching and *flow* of pyrrhotite and chalcopyrite around more rigid nuclei composed of gangue and sphalerite, and by the comminution of pyrrhotite grains with development of subgrains, dislocation planes, pressure lamellae and pressure shadows (Fig. 9d). C planes are spaced and discontinuous, whereas C' planes are located in planes close to the global shear plane. We have also observed boudins and pinch-and-swell structures, including tectonic enclaves of chloritized sedimentary rocks within massive sulphide, piercement structures (Maiden et al., 1986), and isoclinal D_1 and D_2 folds (Fig. 6d, e) where sphalerite and galena are often concentrated within saddles of microfold hinges. In pressure shadows, and other low strain domains, granoblastic polygonal textures with no clear orientation are common. Chalcopyrite, being more ductile than pyrrhotite (Gilligan and Marshall, 1987), shows a more pronounced *flow* than sphalerite and galena.

The massive sulphide lenses are also affected by several generations of late brittle faults (Fig. 2) that locally displace the main orebody up to 50 m. Carbonate-filled veinlets associated with secondary pyrite and marcasite are very likely related to this later brittle faulting event (Fig. 7f).

3.4. Magnetic characteristics of the Draa Sfar ore

A brief study of the magnetic properties of the Draa Sfar orebody was conducted in order to determine the type of pyrrhotite (monoclinic or hexagonal), and its magnetic characteristics. Two types of measurements were taken: 1) absolute magnetic susceptibility and, 2) thermoremanent magnetism, determined by heating the samples to the Curie temperature (700 °C) and then cooling (analyst: Y. Chen ISTO Orléans).

The Draa Sfar ore has an absolute magnetic susceptibility of about 2.10^{-3} SI/g, markedly less than that of ore at Hajar ($5.3.10^{-3}$ SI/g). Thermoremanent magnetic profiles show a loss of magnetic susceptibility above the Curie temperature of pyrrhotite (about 340 °C), which indicates that monoclinic pyrrhotite is the only magnetic mineral within the ore (Fig. 10). The

appearance of a broad second peak in magnetic susceptibility centred around 560 °C suggests the appearance of hematite by partial oxidation of pyrrhotite. Partial oxidation of pyrrhotite explains the intensity decrease of the first pyrrhotite peak on the cooling curve.

4. Lead isotope data and implications

Except for one sample, taken from the gossan at Sidi M'Barek, all analyzed samples are from the pyrrhotite-rich, polymetallic massive sulphide orebodies at Tazakourt and Sidi M'Barek and were collected from drill core or underground workings (Table 3). The lead content of the massive sulphide samples is high, ranging from 329 ppm to more than 4%, indicating their variable galena content and corrections for *in situ* radiogenic enrichment are negligible (μ ratio ranging from 0 to 0.79).

(1) and (2) this study: (1): analyses performed by J.J. Peucat (University of Rennes France), (2): analyses performed by B. Cousens, Carleton University, Sudbury, Canada; ACE: A. Edwards (1989, unpublished data); U, Th and Pb contents have been measured in Guemassa laboratory by REMINEX division valorisation by X-ray fluorescence, and ISTO laboratory Orleans by ICP. Analytical uncertainties see text; py = pyrite, po = pyrrhotite, cp = chalcopyrite, sp = sphalerite, gn = galena. nd: no data.

Analytical procedures are summarized herein. Between 10 and 20 mg of sulphide powder was dissolved in 7N HNO₃. The residue was taken up in 1N HBr for Pb chemistry. Pb was separated in Bio-Rad 10-ml polyethylene columns and Dowex AG1-8X anion resin, using 1N HBR to elute other elements and 6N HCl to elute Pb. Total procedural blanks for Pb are < 250 pg. Samples were loaded onto single Re filaments with H₃PO₄ and silica gel, and were run at filament temperatures of 1175 to 1225 °C. Analysis of USGS Standard BCR-1 yields Pb = 13.56 ppm, U = 1.70 ppm, Th = 5.86 ppm, ²⁰⁶Pb/²⁰⁴Pb = 18.818, ²⁰⁷Pb/²⁰⁴Pb = 15.633, and ²⁰⁸Pb/²⁰⁴Pb = 38.633 (average of 2 runs, see Tatsumoto et al., 1972). All mass spectrometer runs were corrected for fractionation using NIST SRM981. The average ratios measured for SRM981 on the new Triton Mass spectrometer are ²⁰⁶Pb/²⁰⁴Pb = 16.892 ± .010, ²⁰⁷Pb/²⁰⁴Pb = 15.432 ± .013, and ²⁰⁸Pb/²⁰⁴Pb = 36.512 ± .038 (2 S.D.), based on 15 runs between May 2005 and January 2006. The fractionation correction, based on the values of Todt et al., (1984), is + 0.13%/amu.

The lead isotopic compositions of the samples range from 18.27 to 19.54 for ²⁰⁶Pb/²⁰⁴Pb (Table 3, Fig. 11). Samples with least radiogenic signatures (near 18.30 ²⁰⁶Pb/²⁰⁴Pb) are from the more Zn-rich Tazakourt orebody and the upper, Zn-rich sulphide lens at Sidi M'Barek, whereas the most radiogenic samples (²⁰⁶Pb/²⁰⁴Pb = 18.50 to 19.01) are from the more Cu-rich, lower sulphide lens at Sidi M'Barek. The sample taken from the Sidi M'Barek gossan also has a radiogenic signature (²⁰⁶Pb/²⁰⁴Pb = 18.84) that suggests that this gossan developed within a more Cu-rich sulphide lens, an interpretation supported by its higher copper grade. The anomalously radiogenic sample (²⁰⁶Pb/²⁰⁴Pb = 19.54) was taken from a late quartz–pyrite vein; for this single pyritic sample, *in situ* radiogenic enrichment likely occurred (Table 3).

The lead isotope compositions of all data define a mixing trend (Fig. 11). Based on their lead isotopic composition, the Draa Sfar massive sulphide orebodies record the imprint of two different hydrothermal fluids. The first is characterized by a lower radiogenic lead content

($^{206}\text{Pb}/^{204}\text{Pb}$ ratio of 18.28 to 18.30) and this fluid formed the more Zn-rich sulphide lenses at Tazakourt and Sidi M'Barek. The second fluid contained more radiogenic lead ($^{206}\text{Pb}/^{204}\text{Pb}$ ratio of 19.01) and formed the lower, Cu-rich lens at Sidi M'Barek. This second more radiogenic fluid is interpreted to have overprinted the upper, more Zn-rich Sidi M'Barek sulphide lens resulting in intermediate Pb isotopic compositions (18.58 to 18.84 $^{206}\text{Pb}/^{204}\text{Pb}$ ratio). The sample with the highest $^{206}\text{Pb}/^{204}\text{Pb}$ ratio (18.77) from the Sidi M'Barek upper lens is interpreted to be the most influenced by the more radiogenic hydrothermal fluid, although chalcopyrite is not observed in this sample. The Draa Sfar lead isotopic compositions plotted on the growth curve of Stacey and Kramers (1975) are consistent with a continental crustal source for the lead, and by inference the other metals (Zn, Cu, Fe) without a perceptible mantle contribution.

The variation in the lead isotopic composition of the Draa Sfar deposit is most similar to that of the Miocene Kuroko VMS deposits, in which the black, Zn-rich ore and the yellow, Cu-ore have significantly different Pb isotopic compositions within each ore deposit ([Sato, 1975] and [Fehn et al., 1983]). This variability in lead isotopic composition between ore types at Draa Sfar is distinctly different from VMS deposits of the Iberian Pyrite Belt, where isotopic compositions are clustered at both the deposit, and belt scale (Marcoux, 1998).

The magnitude of the variation in lead isotopic compositions at Draa Sfar is significantly larger than that of the Miocene Kuroko VMS deposits (about 0.73 *versus* 0.10 for $^{206}\text{Pb}/^{204}\text{Pb}$ ratios). In the Kuroko deposits, the differences in isotopic composition between black and yellow ores are interpreted to reflect a shift in the proportion of lead contributed from the same sources (Fehn et al., 1983); however, we consider that such a large variation at Draa Sfar indicates the contribution of lead from different sources. Although lead isotopic data for the different host rock types that constitute the regional stratigraphy are not available, the least radiogenic signature ($^{206}\text{Pb}/^{204}\text{Pb} = 18.28$) is similar to that for Visean igneous rocks of the Iberian Pyrite Belt that have been corrected for U decay. This suggests a major, if not unique, contribution of volcanic rocks as a source of metal (Marcoux, 1998). The more radiogenic Cu-ore signature ($^{206}\text{Pb}/^{204}\text{Pb} = 19.01$) is close to the Doe and Zartman (1979) upper crust Pb isotope evolutionary curve. This suggests a crustal sedimentary source and, in particular, a contribution from black shales. The latter is typically highly radiogenic, and is abundant in the lithostratigraphic succession (Sarhlef series) that host the Draa Sfar VMS deposit.

The nearby Hajar VMS has a different, less radiogenic isotopic composition ($^{206}\text{Pb}/^{204}\text{Pb} = 18.17$), close to VMS deposits of the Iberian Pyrite Belt (average of 18.18 for $^{206}\text{Pb}/^{204}\text{Pb}$ ratios). Such a difference between the lead isotopic composition of the sulphide ores at Hajar and Draa Sfar may reflect differences in the metal source and (or) the age of the deposits. However, as the two ore deposits are of Visean age, this alone cannot explain the difference in lead isotopic compositions. In spite of similar mineralogical, petrologic and chronological attributes, the pyrrhotite-rich sulphide ores at Hajar and Draa Sfar may differ in their lead and metal sources.

5. Ar–Ar dating

Two samples (DS 57 and DSF3G), of polycrystalline sericite and strongly sericitized plagioclase respectively, were analysed with a ^{39}Ar – ^{40}Ar laser probe (CO_2 Synrad[®]). Grains of the samples were handpicked under a binocular microscope from crushed rocks (0.3 to 2 mm fraction). Selected grains were wrapped in Al foil to form packets (11 mm × 11 mm × 0.5 mm). These packets were stacked up to form a pile, within which

packets of flux monitors were inserted every 10 samples. The stack, put in an irradiation can, was irradiated at the McMaster reactor (Hamilton, Canada). The irradiation standard was the sanidine TCR-2 (28.34 Ma according to Renne et al., 1998). The sample arrangement allowed us to monitor the flux gradient with a precision of $\pm 0.2\%$.

The step-heating experimental procedure is described in detail in (Ruffet et al., 1995) and (Ruffet et al., 1997). Blanks are performed routinely each first or third run, and are subtracted from the subsequent sample gas fractions. Analyses are performed on a Map215[®] mass spectrometer. To define a plateau age, a minimum of three consecutive steps is required, corresponding to a minimum of 70% of the total $^{39}\text{Ar}_K$ released, and the individual fraction ages should agree to within 1σ or 2σ with the integrated age of the plateau segment. All discussed ^{39}Ar – ^{40}Ar results are displayed at the 1σ level. Pseudo-plateau ages, with less than 70% of $^{39}\text{Ar}_K$ released, can also be defined (Table 4).

The two samples display concordant (1σ level) plateau and pseudo-plateau ages at *ca.* 331 Ma with very distinct age spectrum shapes (Fig. 12). Sample DSF3G displays a flat age spectrum over 90% of total $^{39}\text{Ar}_K$ degassing with an increase from the low to the high temperature steps of the measured $^{37}\text{Ar}_{Ca}/^{39}\text{Ar}_K$ ratios, up to *ca.* 100. The corresponding calculated high CaO/K₂O ratios ($\text{CaO}/\text{K}_2\text{O} = ^{37}\text{Ar}_{Ca}/^{39}\text{Ar}_K \times 2.179$; Deckart et al., 1997), from *ca.* 20 up to *ca.* 210, and the low measured $^{39}\text{Ar}_K$ values suggest that the analysed fragment is Ca-rich/K-poor, and mainly composed of plagioclase.

The other sample DS57 yields a much more disturbed age spectrum, with globally a saddle shape for the last 80% of $^{39}\text{Ar}_K$ degassing, corresponding to a saddle shape of the $^{37}\text{Ar}_{Ca}/^{39}\text{Ar}_K$ spectrum. Excepting the three last steps, the measured $^{37}\text{Ar}_{Ca}/^{39}\text{Ar}_K$ ratios are 2 to 3 orders of magnitude lower than for previous sample and are evidence for degassing of a K-rich phase. The correspondence between the $^{37}\text{Ar}_{Ca}/^{39}\text{Ar}_K$ saddle minimum and the pseudo-plateau age displayed by age spectrum (330.8 ± 1.1 Ma) suggests that degassing of this K-rich phase, probably sericite, is preponderant in this temperature domain.

Despite similar plateau and pseudo-plateau ages, the two analysed samples show very distinct $^{37}\text{Ar}_{Ca}/^{39}\text{Ar}_K$ spectra which characterize a Ca-rich and a K-rich phase, respectively. The age of the sericite must be younger or equal to the age of the plagioclase. This appears to be the case but what is the significance of the age indicated by sample DSF3G? Two possibilities exist: 1) if the 331.7 ± 7.9 Ma age is the age of the plagioclase, then it dates volcanism and suggests that sericitization, a product of VMS hydrothermal alteration, was sub-synchronous with rhyolite emplacement; and 2) if the 331.7 ± 7.9 Ma age is the age of sericite present within plagioclase then this age reflects the age of VMS alteration and not necessarily the age of the rhyolite (or of volcanism).

Based on the Ar–Ar ages, we propose that the sericite, a product of the VMS hydrothermal system responsible for sericitization and the formation of the Draa Sfar deposit, has an age of *ca.* 331 Ma. This age is probably contemporaneous with or close to (slightly younger) the age of rhyolite emplacement and volcanism.

6. Discussion: a genetic model

Estimating the temperature of formation for such a highly deformed, recrystallized and annealed massive sulphide deposit such as Draa Sfar is extremely difficult. The possible depositional equilibrium of various sulphides has been so strongly disturbed by

metamorphism (for which no pressure range can be estimated) that experimental curves for the Fe–Zn–As–S system (Scott, 1983) or the Fe-content of sphalerite and pyrrhotite cannot be used as geothermometers. The chlorite geothermometer (Cathelineau and Nieva, 1985) indicates temperatures ranging from 276 °C to 346 °C (hydrothermal clinochlore associated with the massive sulphides, Belkabit et al., 2008-this issue). Within this temperature range, pyrrhotite and sphalerite compositions (about 11 to 12 mol% FeS for the latter) indicate an fS_2 of 10^{-10} – 10^{-11} ([Kissin and Scott, 1982] and [Scott, 1983]), which is in good agreement with the presence of chalcopyrite and the absence of pyrite and löllingite. Considering all of the above, it seems reasonable to propose that the Draa Sfar ore deposit formed under thermal conditions considered “normal” for Palaeozoic massive sulphides orebodies, *i.e.*, with a pH = 4, NaCl = 1 m, $a_{H_2S} = 10^{-3}$, $fO_2 = 10^{-35}$ atm, and $SO_4^{2-}/H_2S = 10^{-1}$ within a temperature range of 250° to 300 °C (Large, 1992).

The abundance of pyrrhotite is a conspicuous feature of the Draa Sfar deposit and, in general, of all the Hercynian VMS deposits in the Jebilet and Guemassa terranes. A notable exception is the Lachach deposit, which contains abundant pyrite (E. Marcoux, unpublished data). The origin of the pyrrhotite (*i.e.* primary or metamorphic) is uncertain (*e.g.*, Hajar; Ed Debi et al., 1998).

At Draa Sfar, all observations suggest a primary origin for the pyrrhotite. The metamorphic transformation of pyrrhotite to pyrite requires a high temperature that would also have resulted in the development of calc–silicate minerals such as wollastonite, hedenbergite or andradite, which are unknown or present in only trace amounts at Draa Sfar. The mineral assemblages andradite–pyrrhotite and wollastonite–pyrrhotite are stable above 300 °C at pressures of < 2 kbar and where fS_2 is less than 10^{-11} (Gamble, 1982). The low-grade metamorphic assemblages at Draa Sfar (chlorite and absence of biotite) suggest that the temperatures required to transform pyrrhotite to pyrite by metamorphic desulphidation reactions were not achieved. Relict primary “pyrite” has not been observed. Pyrite occurs as secondary porphyroblasts after pyrrhotite including relicts of pyrrhotite incorporated during growth, and also occurs as late crosscutting veins. Slow metamorphic cooling of pyrrhotite is a process accompanied by little change in the composition of this mineral and any sulphur released is rapidly fixed as pyrite (Vaughan and Craig, 1997). Scarcity of secondary pyrite at Draa Sfar indicates only limited release of sulphur during metamorphism.

The abundance of primary pyrrhotite at Draa Sfar and in other VMS deposits of the Jebilet and Guemassa terranes can be ascribed to several processes. Many pyrrhotite- and magnetite-bearing Archean VMS deposits are interpreted to reflect the lower fO_2 (10^{-42} atm. at 300 °C) and a_{H_2S} (10^{-3}) content of the ore fluid ([Large, 1977] and [Large, 1992]). Such sulphur- and oxygen-poor fluids have also been suggested to explain the pyrrhotite (and magnetite)-rich VMS deposits in the highly sedimented, Middle Valley of the modern seafloor (Krasnov et al., 1994). Sharpe and Gemmel (2002) suggested that at 250 °C pyrrhotite is stable over a

wide range of pH (2 to 7.5) and low fO_2 conditions, below 10^{-39} and 10^{-41} for total dissolved sulphur (mostly H_2S) in hydrothermal fluid of 10^{-3} m and 10^{-2} m, respectively, as an explanation for pyrrhotite formation within the magnetite-rich, Gossan Hill VMS deposit. At lower S content, the pyrrhotite field drops to very low fO_2 values (10^{-42} atm) and the presence of clinochlore restricts the pH range between 4.3 and 5.3 (Helgeson, 1969 in Large, 1977).

In summary, we believe that very low fO_2 conditions are the main reason for the formation of primary pyrrhotite at Draa Sfar, and neither a high temperature nor a low aH_2S (below 10^{-3}) is required. The highly anoxic conditions required to stabilize pyrrhotite over pyrite are consistent with the deposition setting and reconstruction of the Draa Sfar deposit, which is interpreted to form within a restricted, sediment-starved, anoxic basin characterized by the deposition of carbonaceous, pelagic sediments along the flank of a rhyolitic flow-dome complex that was buried by pelitic sediments (Belkabit et al., 2008-this issue). Deposition of sulphides likely occurred at and below the seafloor within anoxic and carbonaceous muds as proposed for the Hajar deposit (Leblanc, 1993).

However, between the formation of the initial Zn-rich and subsequent more Cu-rich orebodies the lead isotope data suggest that hydrothermal circulation changed to reflect a different source rock, this source rock may have been richer in copper such as black shales, or the fluid may have had a higher temperature increasing its ability to dissolve and carry copper.

Nevertheless, metamorphism and deformation had a visible and pronounced effect on the pyrrhotite-rich massive sulphides at various scales. The sheet-like form of the orebodies may, in part, be primary but it also certainly reflects accentuation and flattening during deformation. Metamorphism has resulted in sulphide recrystallization and the characteristic development of pyrrhotite with 120° triple junctions. Ductility contrasts during deformation and metamorphism resulted in the development of shear zones along the contacts of the orebodies with their host sedimentary rocks, the development of boudinaged ore lenses, and the development of shear zones that produced local mylonite zones within the orebodies. These metamorphic and deformational induced transformations mask primary textures within the orebodies and have modified metal zoning patterns.

7. Conclusions

The Draa Sfar deposit is characterized by an abundance of primary pyrrhotite, absence of primary pyrite, and by two Zn-rich and one Cu-rich sulphide orebodies, each with its own lead isotope composition. Draa Sfar is a “classical” VMS deposit generated by metal-bearing, acidic hydrothermal fluids derived through the interaction of evolved seawater with sedimentary and lesser volcanic rocks, with convection driven by a high geothermal gradient associated with rifting and localized volcanism *ca.* 331 Ma. A multistage model including; i) precipitation of Zn-rich polymetallic sulphides and ii) later deposition of more Cu-rich ore at higher temperature, also seems valid for Draa Sfar ([Large, 1977] and [Franklin et al., 2005]).

VMS deposits of the Jebilet and Guemassa terranes (Fig. 1a, b) have been compared to those of the Iberian Pyritic Belt (IPB) deposits ([Bernard et al., 1988], [Leistel et al., 1998] and [Lescuyer et al., 1998]). These authors have stressed the similarities between the two districts in term of age, volcanosedimentary host rocks, and paleogeodynamic setting. The Iberian and Moroccan terranes lie in an epicontinental domain of the outer zone of the Hercynian belt and the VMS deposits formed within an environment that has a high sedimentary component, one that is more siliciclastic in the IPB and more pelagic in the Jebilet and Guemassa areas (Lescuyer et al., 1998). Other differences between the IPB and Jebilet VMS deposits include the mineralogy of the ore, with the predominance of pyrrhotite and scarcity of lead in the Jebilet and Guemassa deposits, in contrast to the pyrite-rich and more polymetallic VMS deposits of the IPB. These differences in mineralogy are interpreted to reflect differences in the depositional environment of the sulphide ores, and the differences in lead isotopic characteristics reflect differences in source rocks.

In spite of the diachronous emplacement between the IPB deposits (late Devonian to Visean) and Moroccan deposits (Visean), it is possible to define a main metalliferous peak around 340 ± 10 Ma that occurred during rifting, which followed a Devonian compression event (Lescuyer et al., 1998).

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Figures and Tables

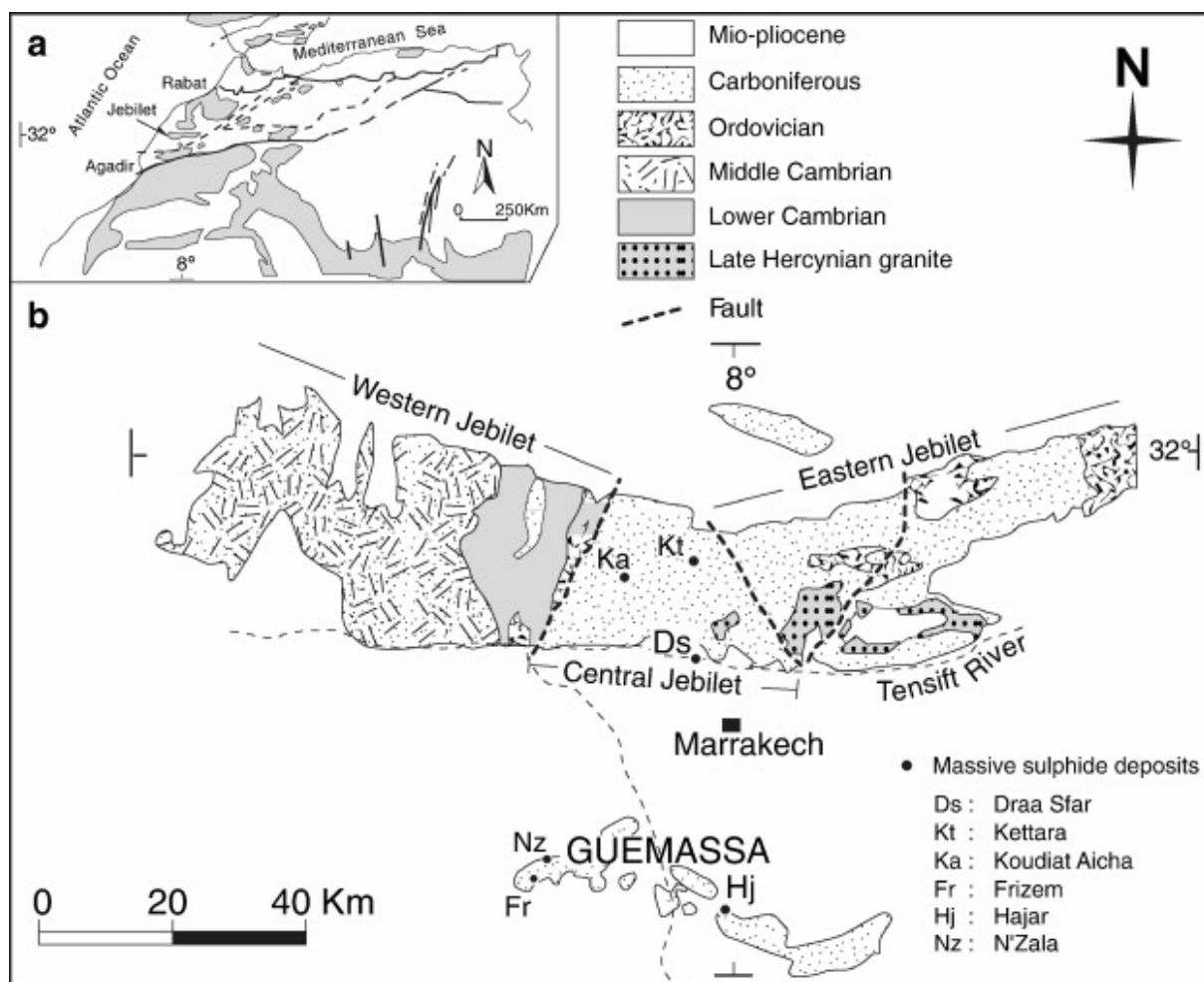


Fig. 1. (a) The location of the Jebilet massif and the distribution of Palaeozoic rocks of North Africa (in grey). (b) General geologic map of the Jebilet and Guemassa Hercynian massifs, around Marrakech, showing the location of the principal massive sulphide deposits (modified from Huvelin, 1977).

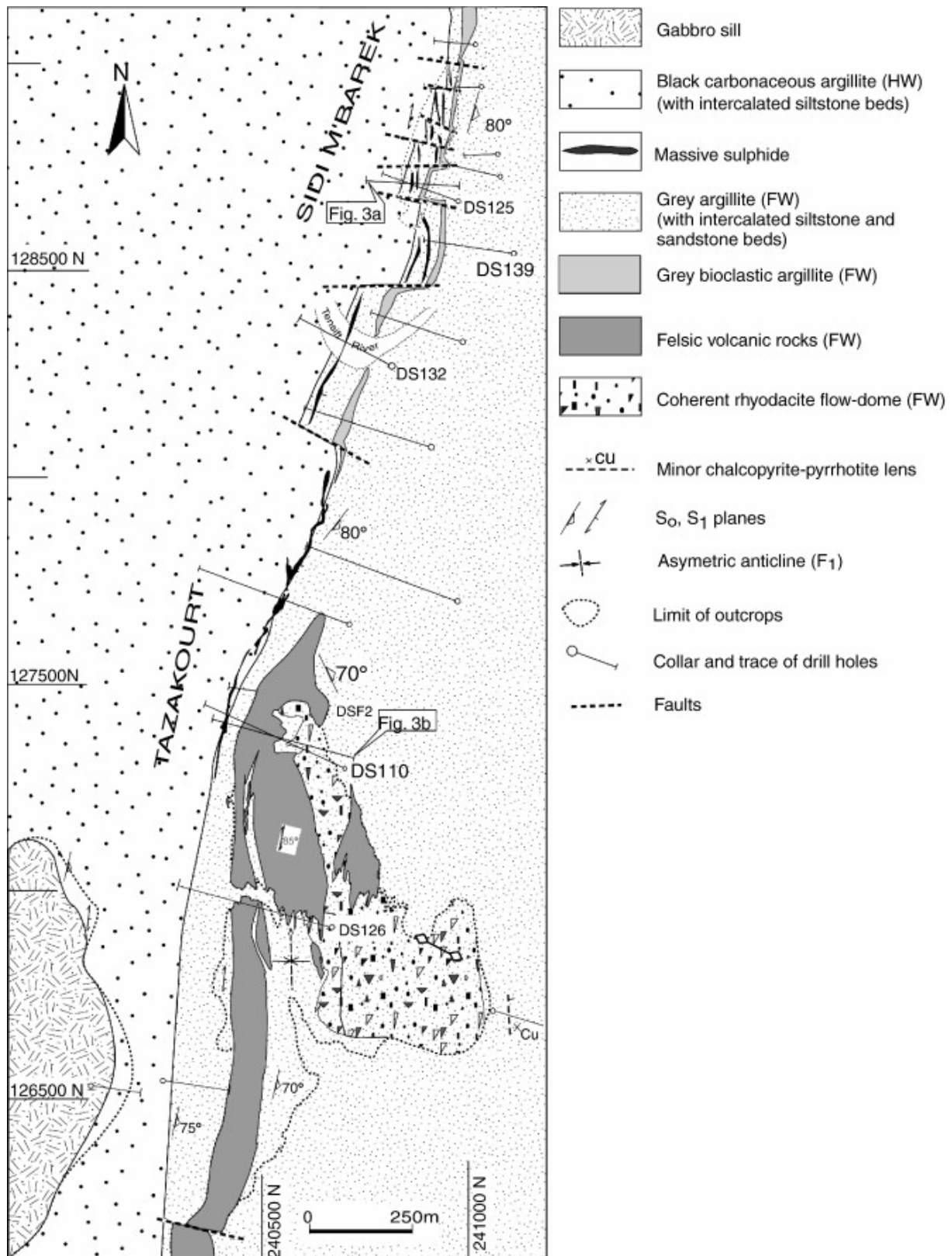


Fig. 2. Geological map of the Draa Sfar area, showing the location of the massive sulphide orebodies, key geologic units and selected diamond drill holes illustrated in Fig. 3 and Fig. 5.

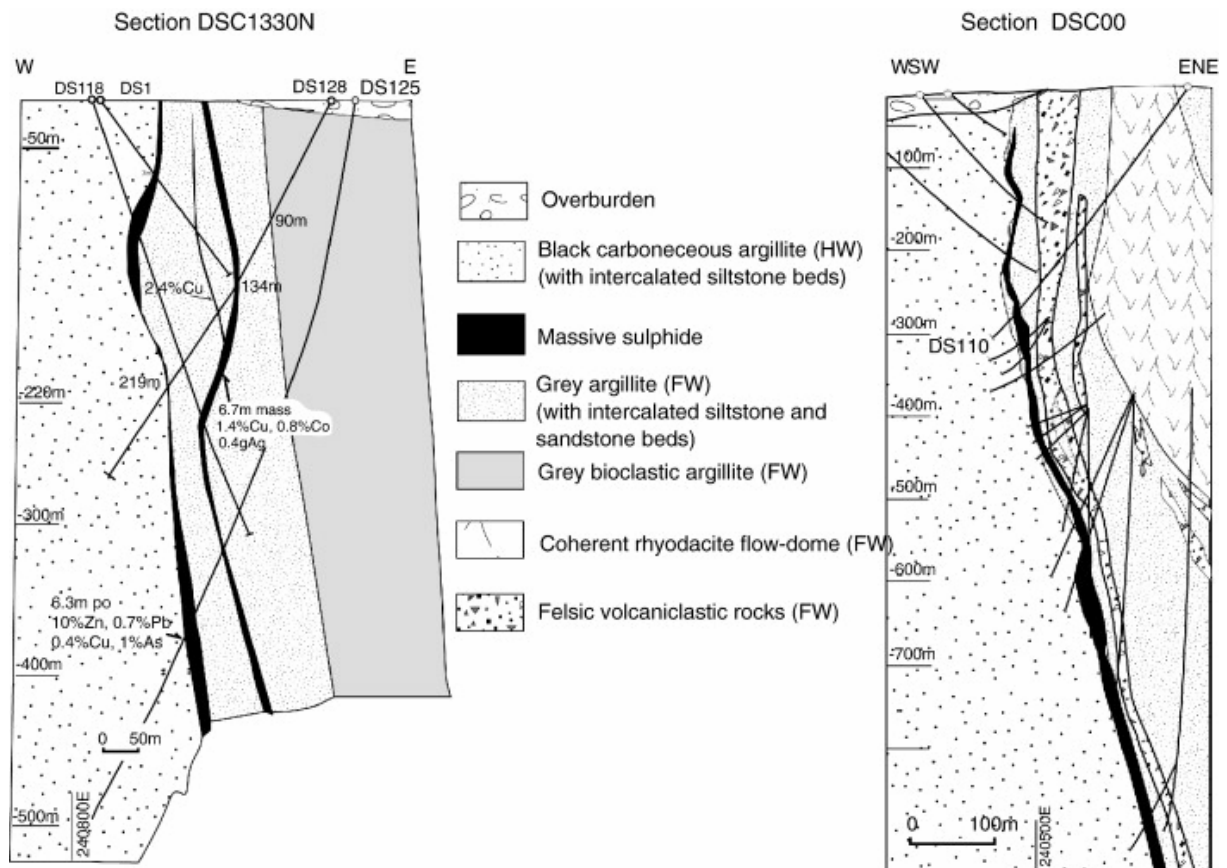


Fig. 3. Vertical east–west cross sections looking north through the Draa Sfar deposit showing the main ore zones and mine stratigraphy. To improve clarity, the mafic dykes and sills are not shown.

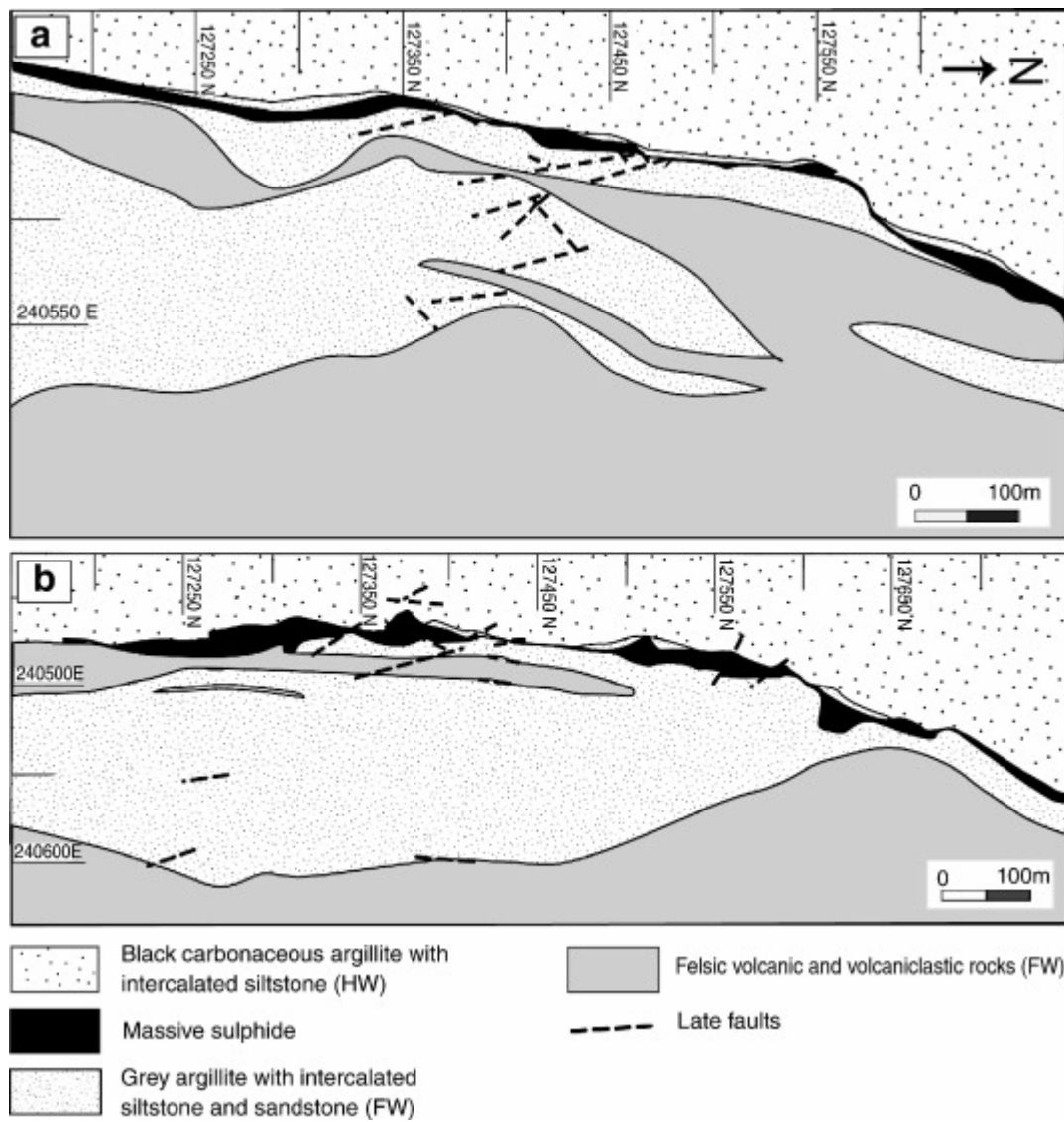


Fig. 4. Simplified geological map of the (a) – 300 and (b) – 400 m levels through the Tazakourt orebody, showing the sheet-like and boudinaged form of the sulphide lens.

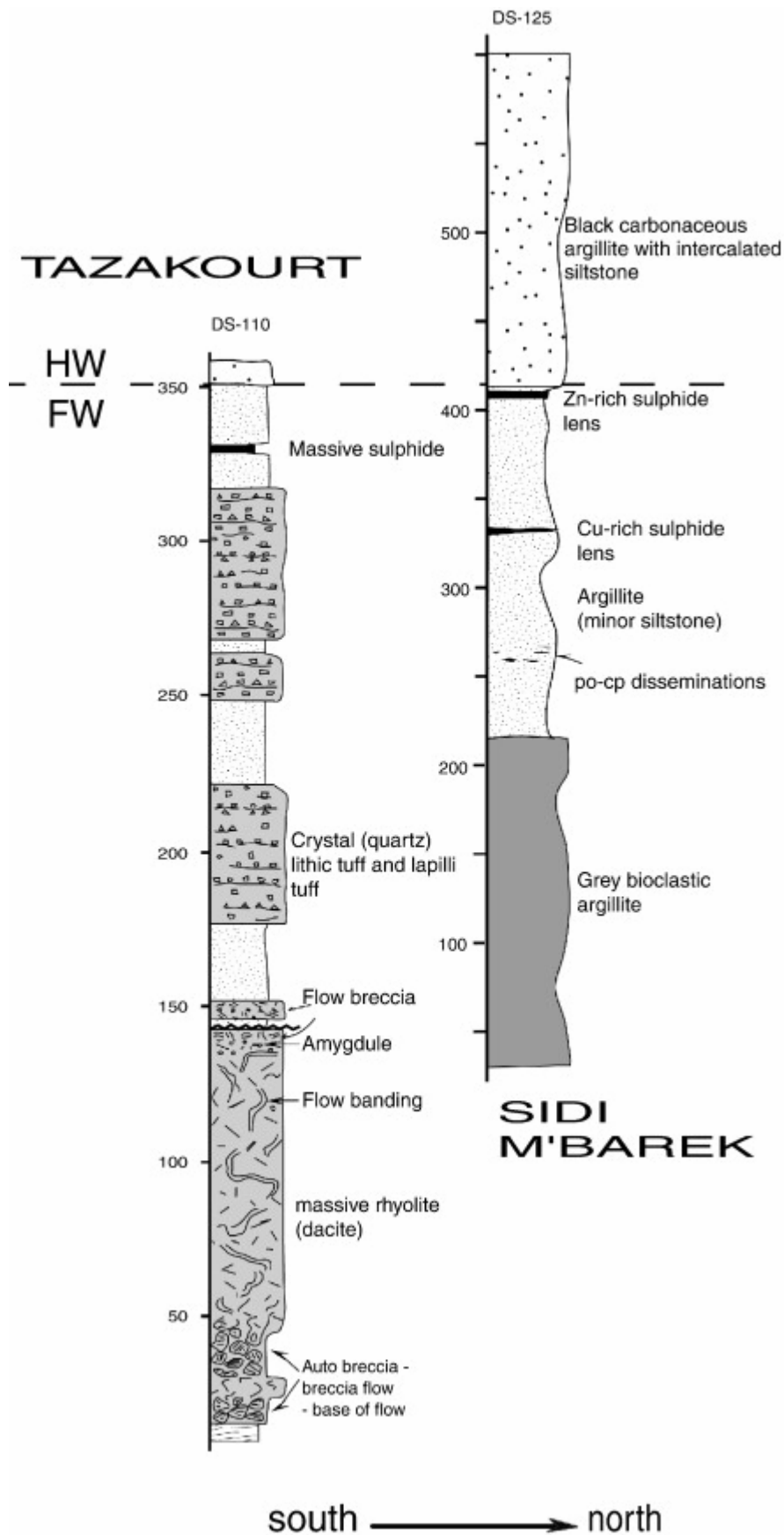


Fig. 5. Restored east-west stratigraphic sections of the Draa Sfar mine stratigraphy, as determined from detail logging of boreholes DS110 and DS125.

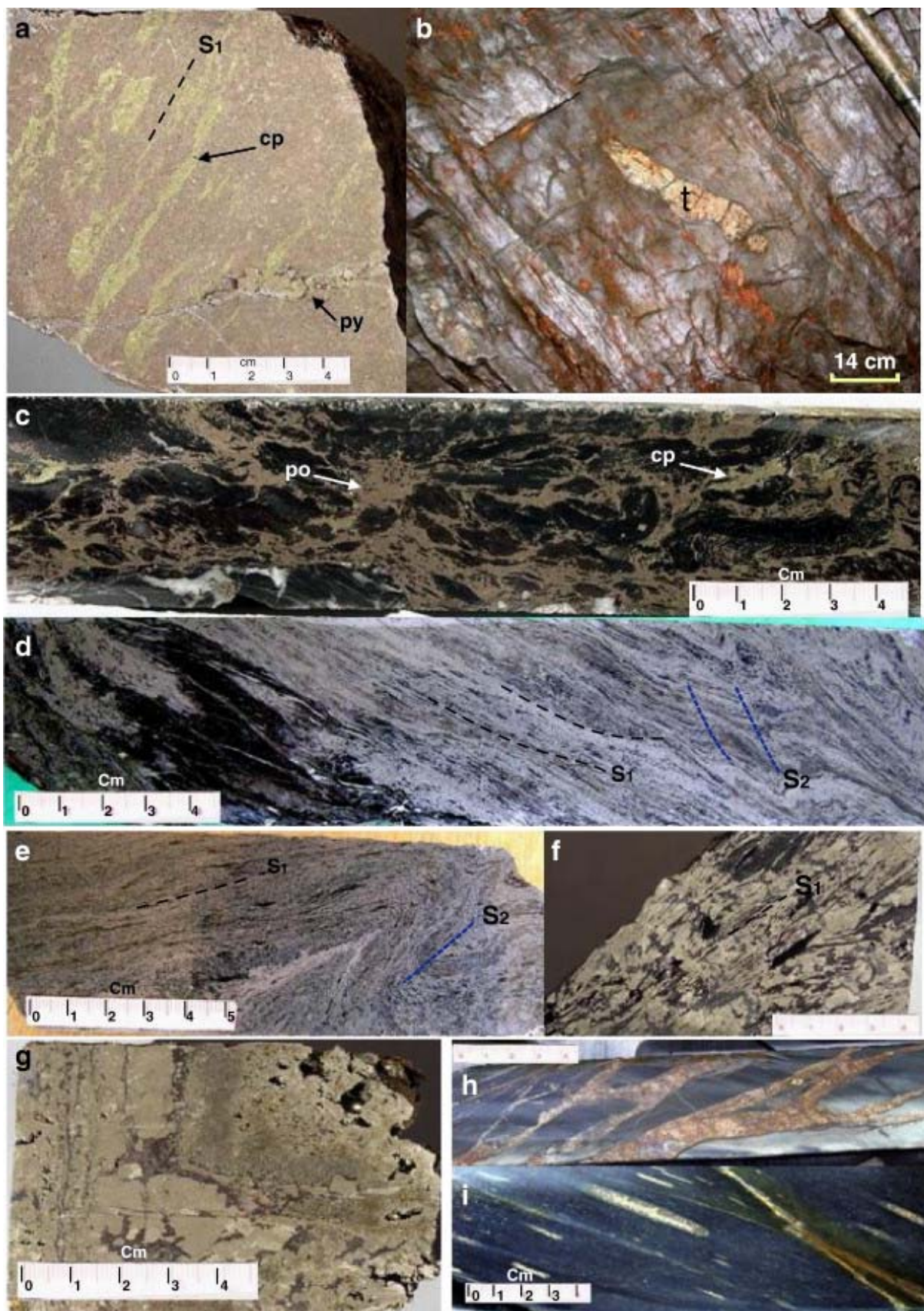


Fig. 6. Photographs of the macroscopic and structural habit of the massive sulphide ore; (a) Polished sample showing the S1 foliation within sulphides marked by the elongation of

chalcopyrite (CP) within pyrrhotite (PO), and a later fracture filled with pyrite (PY). (b) View of the hanging wall of a mine gallery showing a relict of talc–chlorite rimmed by sulphides (t). (c) Polished sample showing ‘durchbewegung’ texture developed within a highly deformed pyrrhotite-rich matrix. Note the differential remobilization, flow, of chalcopyrite (CP) *versus* pyrrhotite (PO). (d), (e) Polished samples illustrating the pronounced S_1 – S_2 foliations and a D1 isoclinal microfold that developed within massive pyrrhotite. (f) Polished sample showing possible pre-tectonic primary pyrite (northern zone of the mine) (g) Polished sample with cataclastic, secondary pyrite (white–yellow) filled by hematite (brown). Polished samples of footwall argillite: (h) pyrrhotite (red–brown) within a fracture network; (i) deformed nodules of disseminated pyrrhotite–chalcopyrite within argillite. Note: Photos scale in cm.

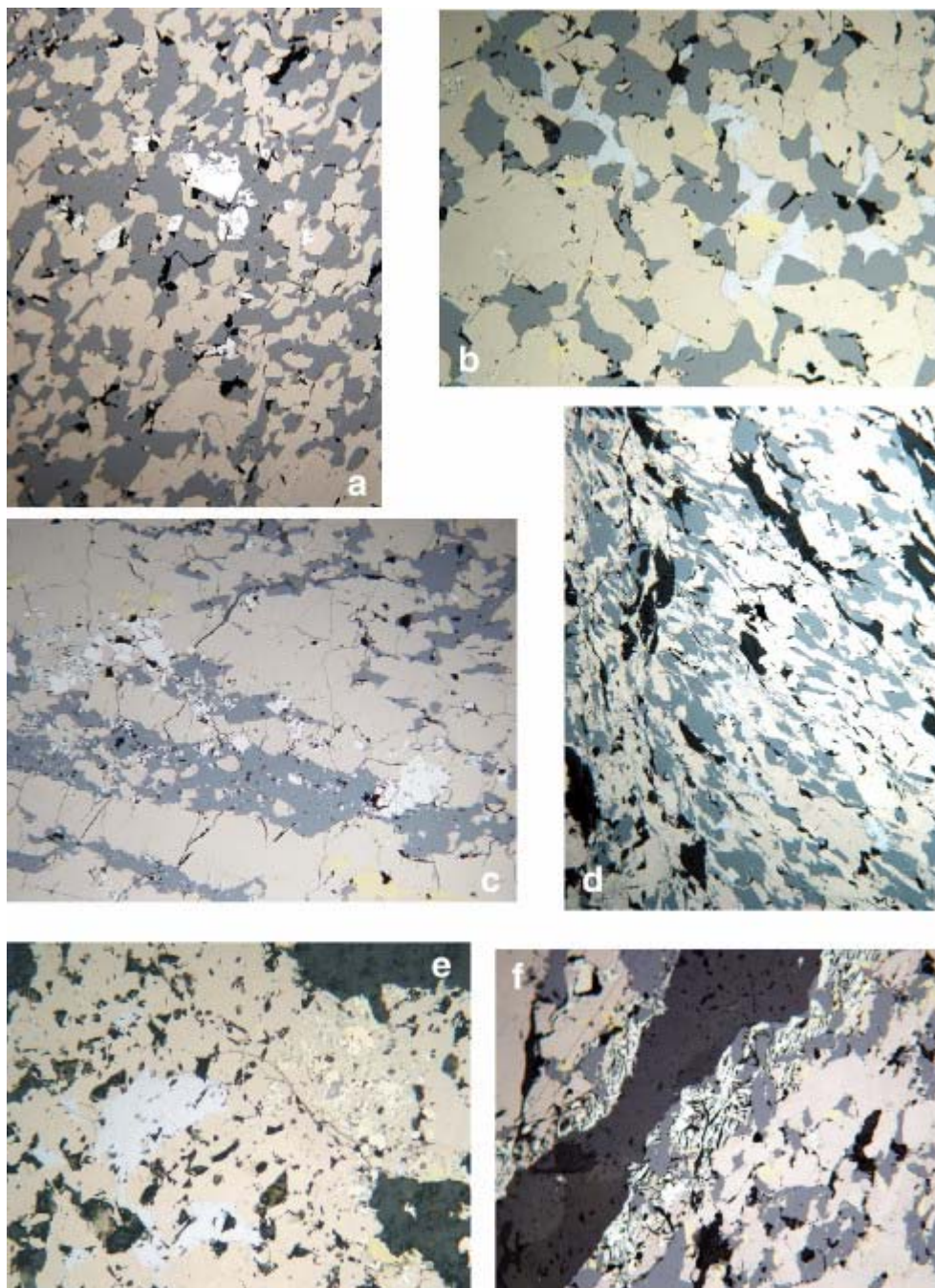


Fig. 7. Reflected, plane-polarized light photomicrographs of Tazakourt orebody. (a) Typical pyrrhotite (pale brown)–sphalerite (gray) ore with disseminated, automorphic arsenopyrite crystals (white) (DS 125, 345 m). Width of field of view: 0.8 mm. (b) Polymetallic ore with pyrrhotite (pale brown), sphalerite (gray), chalcopyrite (yellow) and galena (white) (south crosscut, 120 m). Width of field of view: 0.8 mm. (c) Banded ore with pyrrhotite- and sphalerite-rich bands and accessory arsenopyrite (DDH DS 125, 404 m). Width of field of view: 0.8 mm. (d) Pyrrhotite-early (primary?) pyrite (yellow–white) and sphalerite (gray) ore showing S planes striking NW–SE and C planes striking E–W (DDH DS 110–38, 330 m). Width of field of view: 0.8 mm. (e) Secondary pyrite (pale yellow) and magnetite (gray) in pyrrhotite (DS 125, 404 m). Width of field of view: 0.8 mm. (f) Late carbonate veinlet with an envelope of marcasite (yellow–white) that formed by the replacement of pyrrhotite (south crosscut 120 m). Width of field of view: 0.4 mm.

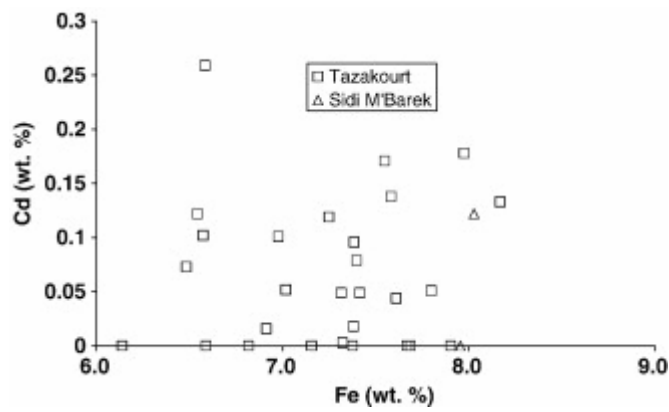


Fig. 8. Bivariate plot of the Fe *versus* Cd content of sphalerite (see Table 2). Electron microprobe analyses, BRGM-ISTO laboratory, Orleans, France.

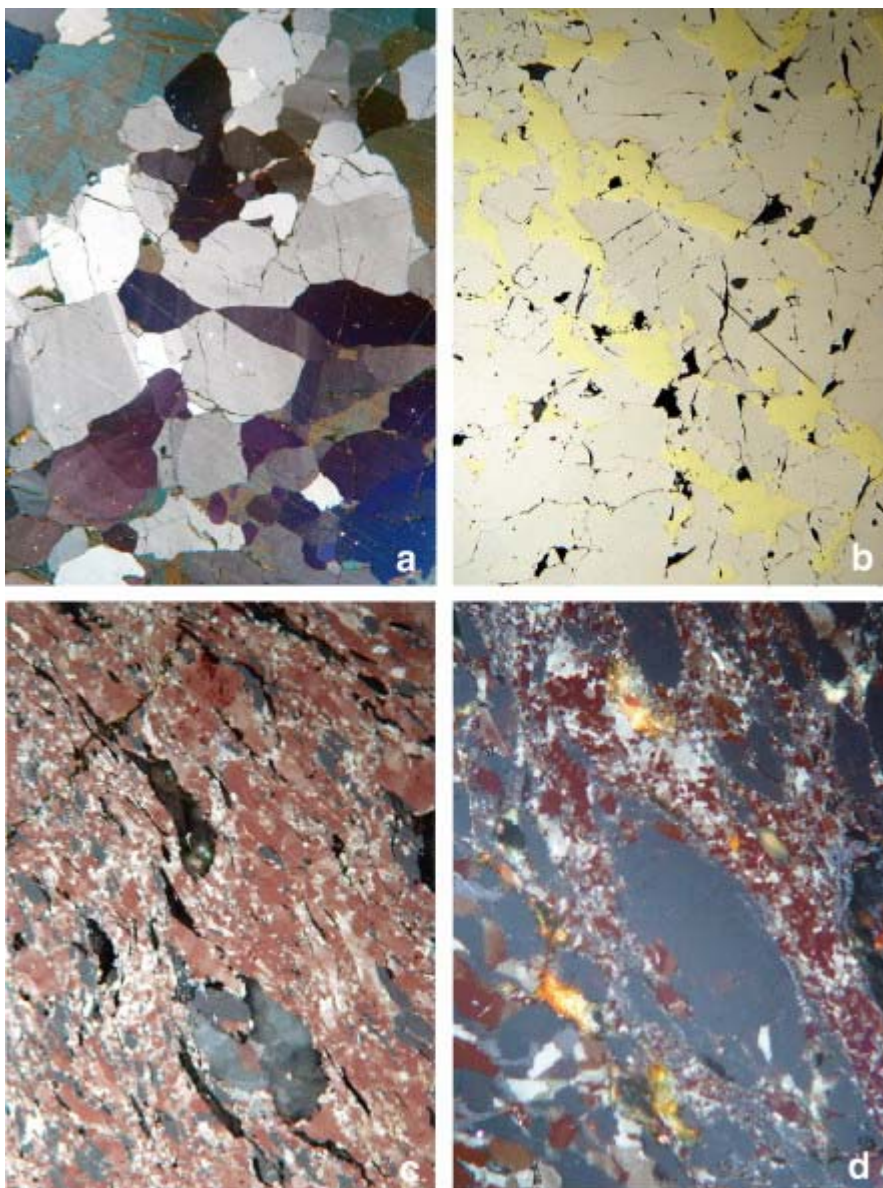


Fig. 9. Reflected light photomicrographs illustrating ore deformation features. (a) Contiguous equigranular pyrrhotite (grey, purple and brown) and chalcopyrite (dark blue) with a

polygonal texture and 120 °C angles. Sidi M'Barek Cu lens DS 128, 139.5 m. Crossed nicols; width of field of view is 1.2 mm. (b) Contiguous equigranular pyrrhotite (light brown) and chalcopyrite (yellow) with a polygonal texture and 120 °C angles. Sidi M'Barek. Cu lens DS 128, 139.5 m. Parallel nicols; width of field of view is 1.2 mm. Photomicrographs a and b are features typical of a metamorphic recrystallization. (c) and (d) mylonitic microfabric in massive sulphide (red brown and white) and quartz (dark grey) characterized by the development of S_1 and S_2 foliations (C, C' and S planes), dislocation planes, pressure lamellae, and pressure shadows imparting a fusiform habit. Tazakourt orebody, DS 110-38, 330 m. Crossed nicols; width of field of view is 0.4 mm for c and d.

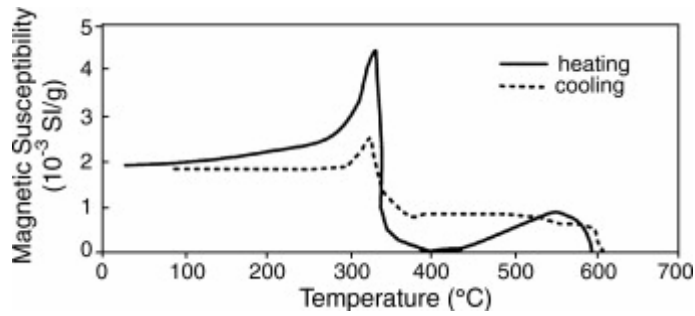


Fig. 10. Thermomagnetic profiles of the pyrrhotite ore.

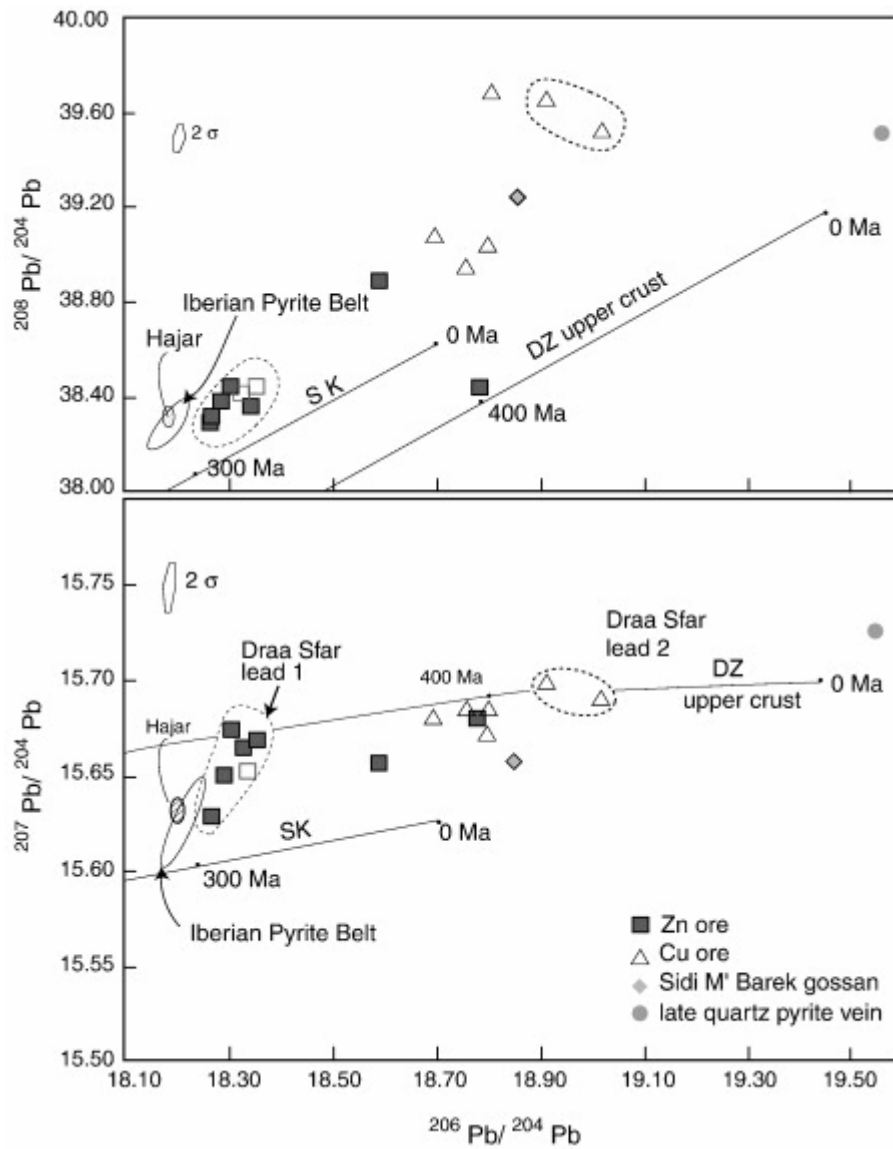


Fig. 11. Lead isotopic compositions of the Draa Sfar ores. Also shown are the lead isotopic compositions for Hajar and for the IPB VMS for comparison (data from Marcoux, 1998). SK: average growth curve of Stacey and Kramers (1975), DZ: upper crust growth curve of Doe and Zartman (1979).

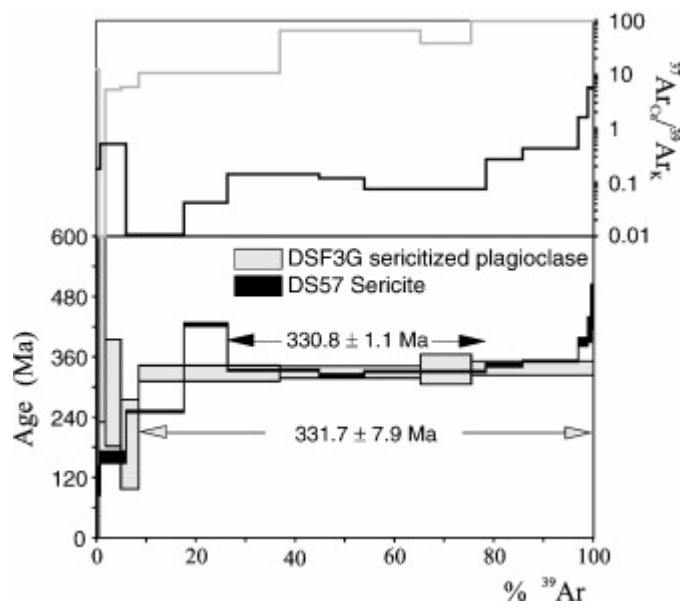


Fig. 12. ^{39}Ar - ^{40}Ar age spectra. The age error bars for each temperature steps are at the 1 level and do not include errors in the J -values. The errors in the J -values are included in the plateau age calculations.

Table 1. : Summary of the main characteristics of the Jebilet and Guemassa VMS deposits

Domain		Jebilet			Guemassa
VMS deposit		Draa Sfar	Kettara	Koudiat Aïcha	Hajar
	Location	15 km NW of Marrakech	32 km NNE of Marrakech	28 km NW of Marrakech	23 km S of Marrakech
	Tonnes (Mt)	10	30	4.5	20
Geology	Grade	5 wt.% Zn, 1.2 wt.% Pb, 0.5 wt.% Cu	55 wt.% Fe, 0.7 wt.% Cu	2.7 wt.% Zn, 1 wt.% Pb, 0.5 wt.% Cu	8 wt.% Zn, 3 wt.% Pb, 0.5 wt.% Cu, 60 g/t Ag
	Main ore minerals	Pyrrhotite, chalcopyrite, sphalerite, arsenopyrite, galena, calcite, quartz, chlorite	Pyrrhotite, pyrite, chalcopyrite, sphalerite, arsenopyrite, galena, calcite, quartz, chlorite	Pyrrhotite, pyrite, chalcopyrite, sphalerite, arsenopyrite, galena, calcite, quartz, chlorite	Pyrrhotite, pyrite, chalcopyrite, sphalerite, arsenopyrite, galena, calcite, quartz, chlorite
	Gossan	Yes	Yes	Yes	Yes
	Host rocks	Volcano-sedimentary	Volcano-sedimentary	Volcano-sedimentary	Volcano-sedimentary
	Deformation	Major	Major	Major	Major
	Age	Visean	Visean		

Table 2.

Electron microprobe analyses of sulphide minerals that constitute the Draa Sfar deposit

	<i>Pyrrhotite</i>															
	Tazak ourt	Tazak ourt	Tazak ourt	Tazak ourt	Tazakou rt	Tazak ourt	Tazak ourt	Tazak ourt	Tazak ourt	Tazak ourt	Tazak ourt	Tazak ourt	Tazak ourt	Tazak ourt	Tazak ourt	
	DS 7	DS 7	DS 7	DS 7	DS 7	DS 7	DS 1	DS 1	DS 1	DS 1	DS 110/37	DS 110/37	DS 110/37	DS2	DS2	
Fe	60.45	60.48	60.03	60.18	60.31	60.40	60.54	59.88	59.68	59.66	60.09	59.77	60.27	59.88	60.02	
S	39.43	39.55	39.32	39.68	39.69	39.84	39.53	39.76	39.52	39.67	39.57	39.76	39.56	39.46	39.73	
As	0.06	0.07	0.06	0.13	0.03	0.11	0.00	0.03	0.03	0.02	0.06	0.09	0.08	0.04	0.04	
Co	0.00	0.00	0.00	0.00	0.00	0.02	0.00	0.00	0.00	0.01	0.00	0.01	0.05	0.02	0.05	
Bi	0.10	0.00	0.03	0.16	0.20	0.13	0.04	0.00	0.12	0.00	0.12	0.21	0.00	0.04	0.13	
Total	100.05	100.10	99.45	100.14	100.23	100.50	100.11	99.66	99.34	99.35	99.84	99.84	99.95	99.43	99.98	
Fe cat	46.73	46.74	46.69	46.49	46.57	46.49	46.79	46.26	46.42	46.32	46.55	46.28	46.62	46.53	46.40	
S cat	53.09	53.22	53.27	53.39	53.37	53.41	53.20	53.51	53.54	53.64	53.39	53.62	53.30	53.41	53.50	
XFeS	0.93	0.93	0.93	0.93	0.93	0.93	0.94	0.93	0.93	0.93	0.93	0.93	0.93	0.93	0.93	
	Tazak ourt	Tazak ourt	Tazak ourt	Tazak ourt	Tazakou rt	Tazak ourt	Tazak ourt	Tazak ourt	Tazak ourt	Tazak ourt	Tazak ourt	Tazak ourt	Tazak ourt	Tazak ourt	Avera ge Tazak ourt	

	DS2	DS 125b	DS 125b	DS 125b	DS 125b	DS 23	DS 23	DS 23	DS 23	DS 23	DS 23	DS 23	DS 22	DS 22		
Fe	60.17	60.71	61.03	60.54	60.50	61.03	60.58	60.56	60.68	60.58	60.66	60.81	60.58	60.18	60.35	
S	39.48	39.64	40.01	40.03	40.07	40.05	39.70	40.08	39.87	39.69	39.98	39.75	39.68	39.67	39.72	
As	0.08	0.10	0.04	0.02	0.03	0.10	0.10	0.00	0.12	0.06	0.05	0.06	0.09	0.05	0.06	
Co	0.00	0.00	0.03	0.01	0.03	0.01	0.18	0.08	0.09	0.00	0.00	0.04	0.05	0.05	0.02	
Bi	0.02	0.03	0.04	0.00	0.00	0.02	0.00	0.20	0.09	0.07	0.00	0.00	0.07	0.13	0.07	
Total	99.74	100.48	101.15	100.60	100.63	101.21	100.56	100.92	100.85	100.40	100.69	100.67	100.47	100.09	100.22	
Fe cat	46.62	46.75	46.67	46.47	46.41	46.63	46.61	46.40	46.57	46.68	46.54	46.72	46.62	46.51	46.55	
S cat	53.28	53.17	53.29	53.51	53.53	53.30	53.21	53.49	53.28	53.27	53.43	53.20	53.19	53.39	53.36	
XFeS	0.93	0.93	0.93	0.93	0.93	0.93	0.93	0.93	0.93	0.93	0.93	0.93	0.93	0.93	0.93	
	Sidi M'Bar ek	Sidi M'Bar ek	Sidi M'Bar ek	Sidi M'Bar ek	Sidi M'Barek	Sidi M'Bar ek	Sidi M'Bar ek	Sidi M'Bar ek	Average Sidi M'Bar ek							
	DS 128	DS 128	DS 128	DS 128	DS 128	DS 128	DS 25	DS 25								
Fe	60.69	60.17	60.88	59.85	60.18	59.90	60.13	60.34	60.27							
S	39.70	39.78	40.08	39.81	39.93	39.62	39.81	39.62	39.79							
As	0.11	0.08	0.09	0.00	0.07	0.03	0.00	0.15	0.07							

Co	0.05	0.11	0.09	0.11	0.07	0.10	0.13	0.08	0.09							
Bi	0.00	0.11	0.05	0.04	0.03	0.01	0.11	0.00	0.04							
Total	100.55	100.24	101.19	99.81	100.27	99.65	100.17	100.19	100.26							
Fe cat	46.66	46.39	46.51	46.27	46.35	46.41	46.38	46.58	46.44							
S cat	53.15	53.42	53.33	53.61	53.56	53.46	53.48	53.27	53.41							
XFeS	0.933	0.928	0.930	0.925	0.927	0.928	0.928	0.932	0.929							
	<i>Arsenopyrite</i>															
	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Average Tazakourt
	DS 7	DS 7	DS 7	DS 110/37	DS 110/37	DS 110/37	DS 110/37	DS 125b	DS 125b	DS 125b	DS 125b	DS 125b	DS 125b	DS 125b	DS 23	
Fe	33.51	33.62	34.06	33.25	33.32	33.58	33.40	33.33	33.56	33.68	33.92	33.53	33.08	33.61	32.95	33.49
S	20.55	20.27	20.58	20.75	20.18	20.14	20.16	19.80	20.13	20.16	20.01	19.91	20.03	19.98	18.56	20.08
As	45.62	45.70	45.88	45.24	45.52	45.59	45.64	46.89	46.78	46.43	46.37	46.75	46.44	46.69	48.24	46.25
Co	0.01	0.01	0.00	0.00	0.00	0.00	0.00	0.09	0.00	0.00	0.03	0.13	0.47	0.18	0.02	0.06
Se	0.03	0.01	0.13	0.00	0.34	0.27	0.26	0.00	0.00	0.00	0.07	0.00	0.00	0.11	0.00	0.08
Total	99.71	99.60	100.64	99.24	99.36	99.58	99.47	100.11	100.47	100.27	100.40	100.31	100.01	100.57	99.77	99.97
As cat	32.27	32.44	32.19	32.08	32.43	32.40	32.45	33.33	33.03	32.81	32.79	33.13	32.95	32.99	34.75	32.80

	<i>Sphalerite</i>															
	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Tazakourt	Average
	DS 7	DS 7	DS 7	DS 7	DS 1	DS 1	DS 1	DS 110/37	DS 110/37	DS 110/37	DS 110/37	DS2	DS2	DS2	DS2	
Fe	7.38	7.16	5.99	6.58	7.80	7.90	7.69	8.17	6.59	6.91	7.25	7.16	7.02	7.67	6.98	7.22
S	32.96	33.12	33.20	33.00	33.17	33.64	33.27	33.25	33.22	33.35	32.96	33.30	33.38	33.09	32.98	33.19
Zn	58.01	59.45	60.87	60.71	58.39	58.63	59.01	58.69	60.11	59.54	59.67	59.37	59.66	56.80	59.85	59.25
Cd	0.10	0.00	0.00	0.26	0.05	0.00	0.00	0.13	0.00	0.02	0.12	0.00	0.05	0.00	0.10	0.06
Cu	1.57	0.21	0.16	0.12	0.22	0.00	0.07	0.11	0.17	0.14	0.01	0.00	0.26	1.95	0.00	0.33
Total	100.02	99.94	100.23	100.67	99.62	100.17	100.03	100.35	100.09	99.96	100.00	99.83	100.37	99.51	99.91	100.05
mole % FeS	11.84	11.27	9.42	10.27	12.36	12.46	12.09	12.81	10.37	10.92	11.37	11.29	11.05	12.47	10.96	11.19
	Sidi M'Bar ek	Sidi M'Bar ek	Sidi M'Bar ek	Sidi M'Bar ek	Tazakourt DS 125b	Tazakourt DS 125b	Tazakourt DS 125b	Tazakourt DS 125b	Tazakourt DS 23	Tazakourt DS 23	Tazakourt DS 23	Tazakourt DS 23	Tazakourt DS 23	Tazakourt DS 22	Tazakourt DS 22	Average
	DS 128	DS 128	DS 128	DS 128												
Fe	8.03	7.96	7.94	7.81	7.41	7.55	7.38	7.40	7.58	7.98	7.61	7.32	6.48	6.54	7.50	
S	33.51	33.01	33.34	33.21	33.48	33.57	33.54	33.46	33.87	33.45	33.22	33.45	33.44	33.54	33.43	
Zn	58.77	58.28	59.77	59.46	60.16	59.94	60.39	60.16	59.50	59.56	60.27	59.89	60.07	60.36	59.75	
Cd	0.00	0.00	0.18	0.00	0.05	0.17	0.00	0.08	0.14	0.18	0.04	0.05	0.07	0.12	0.08	

Cu	0.55	0.68	0.36	0.40	0.08	0.00	0.04	0.06	0.16	0.07	0.00	0.00	0.05	0.00	0.17	
Total	100.86	99.92	101.59	100.88	101.18	101.22	101.35	101.16	101.25	101.23	101.14	100.70	100.12	100.56	100.94	
mole % FeS	12.61	12.60	12.30	12.18	11.51	11.74	11.42	11.49	11.86	12.39						

Table 3.

Lead isotope compositions of Draa Sfar ores

Analyses	Location	Sample	Isotopic ratios			Pb	U	Th	μ	W
			$^{206}\text{Pb}/^{204}\text{Pb}$	$^{207}\text{Pb}/^{204}\text{Pb}$	$^{208}\text{Pb}/^{204}\text{Pb}$	(ppm)	(ppm)	(ppm)	($^{238}\text{U}/^{204}\text{Pb}$)	($^{232}\text{Th}/^{204}\text{Pb}$)
ACE3216	Draa Sfar — drill-hole DDH 18 (95–98 m)	Massive po with small amounts of sp and gn (Zn lens)	18.262	15.630	38.293	46300			# 0	# 0
ACE3215	Draa Sfar (north mine waste)	Massive po with small amounts of sp and gn (Zn lens)	18.263	15.630	38.298	38000			# 0	# 0
(1)	Tazakourt DS 124, 692.8 m	Zn lens — polymetallic sulphides	18.266	15.630	38.316	500	< 4	21	0.51	2.77
(1)	Sidi M'Barek — DS 125a, 345.7 m	Zn lens — po–sp–(gn) polymetallic ore	18.586	15.657	38.891	683	< 4	8	0.38	0.78
(1)	Sidi M'Barek — DS 125b, 404.5 m	Zn lens — po–sp–(gn) polymetallic ore	18.285	15.651	38.380	3788	< 4	9	0.07	0.16
(2)	Sidi M'Barek — DS-125b, 407.3 m	Zn lens — banded polymetallic ore	18.301	15.675	38.440	7200	nd	nd	# 0	# 0
(2)	Sidi M'Barek — DS-139-E5, 600 m	Zn lens — massive ore	18.325	15.665	38.413	856	nd	nd		

Analyses	Location	Sample	Isotopic ratios			Pb	U	Th	μ	W
			$^{206}\text{Pb}/^{204}\text{Pb}$	$^{207}\text{Pb}/^{204}\text{Pb}$	$^{208}\text{Pb}/^{204}\text{Pb}$	(ppm)	(ppm)	(ppm)	($^{238}\text{U}/^{204}\text{Pb}$)	($^{232}\text{Th}/^{204}\text{Pb}$)
(2)	Sidi M'Barek — DS-139-E6, 601 m	Zn lens — massive ore	18.776	15.681	38.441	942	nd	nd		
(2)	Sidi M'Barek — DS-139-E7, 602.2m	Zn lens — massive ore	18.351	15.669	38.449	15620	nd	nd		
(2)	Sidi M'Barek — DS-139-E7, 603m	Zn lens — massive ore	18.334	15.653	38.377	11253	nd	nd		
(2)	Sidi M'Barek — DS-128-1, 133.7 m	Cu lens — contact massive sulphides and py vein	18.797	15.687	39.702	436	nd	nd		
(2)	Sidi M'Barek — DS-129-M2, 401 m	Cu lens — massive po with cp (0,5% Cu)	18.905	15.701	39.670	684	nd	nd		
(2)	Sidi M'Barek — DS-139-E1, 430.45 m	Cu lens — massive ore	18.749	15.686	38.959	1754	nd	nd		
(2)	Sidi M'Barek — DS-139-E3, 432.85 m	Cu lens — massive ore	19.010	15.691	39.541	1557	nd	nd		
(2)	Sidi M'Barek — DS-139-E4, 433.3 m	Cu lens — cp stringers in chlorite matrix	18.687	15.681	39.091	393	nd	nd		
(1)	Sidi M'Barek — DS 128, 139.5 m	Cu lens — po and cp recrystallized ore	18.790	15.673	39.051	329	< 4	8	0.78	1.62
(1)	Gossan Sidi M'Barek — DSR 1	Hematite-goethite gossan	18.846	15.658	39.255	447	< 4	10	0.58	1.49
(2)	Sidi M'Barek — DS-128-M12, 133.5 m	Quartz-py vein	19.543	15.727	39.526	877	nd	nd		

Table 4. : $^{40}\text{Ar}/^{39}\text{Ar}$ analytical data

Step	$^{40}\text{Ar}_{\text{Atm.}}$ (%)	^{39}ArK (%)	$^{37}\text{ArCa}/^{39}\text{ArK}$	$^{40}\text{Ar} / ^{39}\text{ArK}$	Age (Ma)
<i>DSF3G</i>					
1	95.1	1.3	0.00	49.74	427.9 ± 197.7
2	82.3	3.1	5.26	32.25	288.7 ± 106.2
3	97.6	3.6	5.83	20.22	186.4 ± 89.0
4	88.3	28.4	10.60	36.94	327.1 ± 15.8
5	67.8	28.3	64.36	37.26	329.7 ± 12.0
6	43.9	10.2	38.18	37.97	335.4 ± 29.6
Fusion	65.6	24.6	97.89	38.25	337.7 ± 14.0
<i>DS57</i>					
1	74.9	0.8	0.18	12.05	113.3 ± 28.6
2	5.9	5.2	0.51	17.32	160.8 ± 11.7
3	4.9	11.7	0.01	27.80	251.5 ± 1.9
4	2.2	8.7	0.04	49.23	424.0 ± 3.4
5	4.6	18.5	0.14	37.77	333.8 ± 1.6
6	3.4	9.0	0.12	36.51	323.6 ± 2.7
7	3.0	24.5	0.08	37.42	331.0 ± 1.6
8	3.7	7.4	0.27	38.99	343.6 ± 3.5
9	2.2	11.2	0.43	39.97	351.5 ± 2.1
10	0.0	1.9	1.60	44.81	389.7 ± 8.5
11	2.6	0.7	5.77	47.96	414.3 ± 24.7
Fusion	40.0	0.5	5.48	53.94	459.8 ± 44.5

$^{40}\text{Ar}_{\text{atm}}$ = atmospheric ^{40}Ar . $^{40}\text{Ar}^*$ = radiogenic ^{40}Ar . Ca = produced by Ca-neutron interferences. K = produced by K-neutron interferences. Age (Ma) = the date is calculated using the decay constants recommended by Steiger and Jäger (1977). The errors are at the 1st level and do not include the error in the value of the J parameter. Correction factors for interfering isotopes produced by neutron irradiation in the McMaster reactor were ($^{39}\text{Ar}/^{37}\text{Ar}$) Ca = 7.06×10^{-4} , ($^{36}\text{Ar}/^{37}\text{Ar}$) Ca = 2.79×10^{-4} , ($^{40}\text{Ar}/^{39}\text{Ar}$) K = 2.97×10^{-2} .